



# Toxicological profile for Ethylene-vinyl acetate copolymer

***This ingredient has been assessed to determine potential human health effects for the consumer. It was considered not to increase the inherent toxicity of the product and thus is acceptable under conditions of intended use.***

## 1. Name of substance and physico-chemical properties

### 1.1. IUPAC systematic name

Ethene;ethenyl acetate (PubChem)

### 1.2. Synonyms

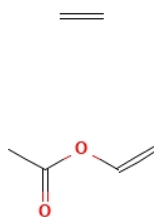
ethene;ethenyl acetate; Acetic acid ethenyl ester, polymer with ethene; Ethylene/vinyl acetate copolymer; ETHYLENE; VINYL ACETATE; 3-Butenoic acid - ethylene (1:1); 3-Butenoic acid, compd. with ethene (1:1); ETHYLENE-VINYL ACETATE; ethylene vinylacetate; Ethylene-vinyl acetate copolymer; Ethene, polymer with ethenyl acetate; (PubChem; ChemSpider)

### 1.3. Molecular formula

C<sub>6</sub>H<sub>10</sub>O<sub>2</sub> (PubChem)

### 1.4. Structural Formula

(PubChem)



### 1.5. Molecular weight (g/mol)

(114.14)n. The copolymer has a weight averaged molecular weight (M<sub>w</sub>) above 6,000 Da, a number averaged molecular weight (M<sub>n</sub>) above 2,000 Da, and a molecular mass range of 200 – 10,000 Da (EFSA, 2014); 114.14 (PubChem)

### 1.6. CAS registration number

24937-78-8

### 1.7. Properties

#### 1.7.1. Melting point

(°C): No data available to us at this time.

#### 1.7.2. Boiling point

(°C): No data available to us at this time.

#### 1.7.3. Solubility

Negligible

#### 1.7.4. *pKa*

No data available to us at this time.

#### 1.7.5. *Flashpoint*

(°C): No data available to us at this time.

#### 1.7.6. *Flammability limits (vol/vol%)*

No data available to us at this time.

#### 1.7.7. *(Auto)ignition temperature*

(°C): No data available to us at this time.

#### 1.7.8. *Decomposition temperature*

(°C): Starts decomposing at >230 (EFSA, 2014)

#### 1.7.9. *Stability*

Stable/Hazardous polymerization will not occur.

#### 1.7.10. *Vapor pressure*

No data available to us at this time.

#### 1.7.11. *log Kow*

No data available to us at this time.

## 2. **General information**

### 2.1. *Exposure*

**Migration of Irganox 1010 from ethylene-vinyl acetate films to foods and food-simulating liquids (Abstract).** In a series of experiments on the migration of the antioxidant Irganox 1010 from ethylene-vinyl acetate (EVA) films into food-simulating liquids and foods, the antioxidant was found to migrate rapidly from EVA film into n-heptane, 100% ethanol and corn oil. The rate of migration into these media was greater from EVA than from low-density polyethylene (LDPE) under comparable conditions. In contrast, little migration of Irganox 1010 was recorded on exposure of the EVA film to aqueous media, whereas migration from LDPE into such media was relatively high.

As taken from Schwöpe AD et al. Food Chem Toxicol. 1987, Apr; 25(4):327-30. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/3583159>

Ethylene/VA copolymer is used as a binding, emulsion stabilising and film forming agent in cosmetics in the EU. As taken from CosIng, undated.

Ethyl vinyl acetate copolymer (CAS RN 24937-78-8) is included on the International Fragrance Association's list of ingredients reported as used in fragrance materials (IFRA).

"Used in orthotics, surfboard and skimboard traction pads, to make artificial flowers, to enhance grip in plastic wraps, improve cold flow of diesel fuel, in HEPA filters, and to make thermoplastic mouthguards and vaginal contraceptives; [REPROTOX] Used for plastic film, laminating, molding,

and coating; [Westlake Polymers MSDS] Permitted for use as an inert ingredient in non-food pesticide products [EPA].”

“**Industrial processes with risk of exposure:** Plastic composites manufacturing.”

As taken from Haz-Map (2020), available at <https://haz-map.com/>

Ethylene/vinyl acetate co-polymer (CAS RN 24937-78-8) is listed as an ingredient (at given concentrations, where specified) in home maintenance (up to 100%), inside the home (10-30%) and personal care products by the CPID.

Ethylene/VA copolymer is used as a topical adhesive, binder, dispersing agent, and film former, and ethylene vinyl acetate as a delivery system in non-medicinal natural health products (both CAS RN 24937-78-8) (Health Canada, 2021).

## *2.2. Combustion products*

### **Polymer degradation and stability**

A Purser furnace has been used to investigate the combustion toxicity of ethylene-vinyl acetate copolymer (EVA) with and without fire retardants, under different fire conditions. Steady state flaming combustion has been studied at equivalence ratios  $\Phi$  varying from 0.5 to 1.5 by driving the materials through the furnace at 750 °C. Yields of CO and CO<sub>2</sub> for EVA containing 27% vinyl acetate, and its fire retarded composites, containing fire retardant fillers are presented. The materials contained 30% EVA and 70% hydrated aluminium oxide (ATH), or 65% ATH and 5% zinc hydroxystannate (ZS), or 5% magnesium borate (MgB) or 5% zinc borate (ZB). In each case the same mass of EVA was used in the determination. The yields of CO per g of polymer from the EVA-fire retardant composite samples showed very similar yields of CO under well ventilated conditions to the pure EVA, but generally higher CO yields than the base polymer under the most toxic fuel rich conditions. The exception to this was the sample containing ATH and zinc borate, which did not take up all the available oxygen under fuel rich conditions, and gave a much lower CO yield, corresponding to an eight-fold reduction in the combustion toxicity. Under fuel rich conditions for EVA, 60% of the carbon was lost as volatile organic species other than CO and CO<sub>2</sub>. For the sample containing zinc borate, this was 50% and for the remaining samples it varied from 20 to 38%. Evidence is presented which indicates that organic material trapped in the solid alumina residue is oxidised to CO, except in the presence of zinc borate, when it appears to be lost as organic carbon.

As taken from HULL T. Richard et al. Polymer degradation and stability ISSN 0141-3910 CODEN PDSTDW . Congrès, European Conference on Fire Retardant Polymers N°8, Alessandria, ITALIE (06/2001) 2002, vol. 77, n°2 (169 p.) (11 ref.), pp. 235-242

## *2.3. Ingredient(s) from which it originates*

No data available to us at this time.

## **3. Status in legislation and other official guidance**

Ethylene-vinyl acetate appears in the latest synoptic list of monomers/additives for use in food-contact materials. The EU Scientific Committee on Food has not given an ADI figure but stated that polymers with a molecular weight above 1000 daltons are very unlikely to be absorbed from the gastrointestinal tract and thus considered not to present a toxicological risk [from use in food packaging] (Commission 2005).

Poly(ethylene-co-vinyl acetate) (CAS RN 24937-78-8) is included on the US FDA list of Indirect Additives used in Food Contact Substances and is covered under Title 21 of the US Code of Federal Regulations (21 CFR), sections 175.105 (Adhesives), 175.300 (Resinous and polymeric

coatings), 176.180 (Components of paper and paperboard in contact with dry food), 177.1200 (Cellophane), 177.1210 (Closures with sealing gaskets for food containers), 177.1350 (Ethylene-vinyl acetate copolymers), 177.1390 (Laminate structures for use at temperatures of 250 °F and above), 178.1005 (Hydrogen peroxide solution) and 179.45 (Packaging materials for use during the irradiation of prepackaged foods).

As taken from FDA, 2023a,c

Copolymer of ethylene and vinyl acetate is listed in the US EPA InertFinder Database (2023) as approved for food and non-food use pesticide products. For food use, vinyl acetate-ethylene copolymer, minimum number average molecular weight (in amu), 69,000 (CAS 24937-78-8) is regulated under Code of Federal Regulations Title 40, Protection of Environment; Part 180, Tolerances and exemptions for pesticide chemical residues in food; Section 180.960 - Polymers; exemptions from the requirement of a tolerance (US EPA, 2023).

“The CEF Panel concluded that the substance ethylene-vinyl acetate copolymer wax does not represent a safety concern for the consumer if the substance is only to be used as an additive up to 2% w/w in polyolefin materials and articles and the migration of low molecular weight oligomeric fraction below 1 000 Da does not exceed 5 mg/kg food.”

As taken from EFSA, 2014.

There is a NONS (Notification of New Substances) dossier on acetic acid, ethenyl ester, copolymer with ethene (F-94; CAS 24937-78-8; EC no. 429-840-1) but, to-date, industry have not submitted a REACH-compliant registration dossier for this chemical (ECHA, undated a).

Acetic acid ethenyl ester, polymer with ethene (poly(ethylene-co-vinyl acetate); CAS RN 24937-78-8; EC no. 607-457-0) is not registered under REACH (ECHA, undated b).

Poly(ethylene-co-vinyl acetate) (CAS RN 24937-78-8; EC no. 607-457-0) is not classified for packaging and labelling under Regulation (EC) No. 1272/2008 (ECHA, 2023).

Ethylene vinyl acetate polymer is listed in the US EPA Toxic Substances Control Act (TSCA) inventory and also in the US EPA 2020 CDR and 2020 CDR Full Exempt lists (Chemical Data Reporting Rule).

#### US EPA Substance Registry Services (SRS)

Ethylene-vinyl acetate copolymer (28 or 9% vinyl acetate) and ethylene-vinyl acetate copolymers are included on the US FDA's list of inactive ingredients for approved drug products. They are permitted for use as ingredients in various products, at the following maximum potencies per unit dose and maximum daily exposures:

Inactive Ingredient	Route	Dosage Form	CAS Number	UNII	Maximum Potency per unit dose	Maximum Daily Exposure (MDE)
ETHYLENE-VINYL ACETATE COPOLYMER (28% VINYL ACETATE)	SUBCUTANEOUS	IMPLANT	24937788	8ILA5X28VS		43 mg

ETHYLENE-VINYL ACETATE COPOLYMER (28% VINYL ACETATE)	VAGINAL	INSERT	2493778 8	8ILA5X28VS		1677 mg
ETHYLENE-VINYL ACETATE COPOLYMER (28% VINYL ACETATE)	VAGINAL	RING	2493778 8	8ILA5X28VS		1686 mg
ETHYLENE-VINYL ACETATE COPOLYMER (9% VINYLACETATE )	VAGINAL	INSERT	2493778 8	4OKC630HS 6		197 mg
ETHYLENE-VINYL ACETATE COPOLYMER (9% VINYLACETATE )	VAGINAL	RING	2493778 8	4OKC630HS 6		250 mg
ETHYLENE-VINYL ACETATE COPOLYMERS	INTRAUTERINE	SUPPOSITOR Y, EXTENDED RELEASE	2493778 8	NA	160 mg	
ETHYLENE-VINYL ACETATE COPOLYMERS	OPHTHALMIC	INSERT, EXTENDED RELEASE	2493778 8	NA	14 mg	
ETHYLENE-VINYL ACETATE COPOLYMERS	OPHTHALMIC	SOLUTION	2493778 8	NA	NA	
ETHYLENE-VINYL ACETATE COPOLYMERS	SUBCUTANEOU S	IMPLANT	2493778 8	NA		61 mg
ETHYLENE-VINYL ACETATE COPOLYMERS	TRANSDERMAL	FILM	2493778 8	NA		12 mg

ETHYLENE-VINYL ACETATE COPOLYMERS	TRANSDERMAL	FILM, EXTENDED RELEASE	2493778 8	NA	735 mg	
-----------------------------------	-------------	------------------------	--------------	----	--------	--

As taken from FDA, 2023c

Acetic acid, ethenyl ester, copolymer with ethene (CAS RN 24937-78-8) is a “polymer identified as low concern to human health by application of expert validated rules” and “poses no unreasonable risk to human health based on Tier I assessment under the NICNAS IMAP assessment framework” (AICIS, 2012)

Acetic acid, ethenyl ester, copolymer with ethene (CAS RN 24937-78-8) is listed on Australian Inventory of Industrial Chemicals (AICIS, formerly NICNAS). As taken from AICIS, undated.

#### **4. Metabolism/Pharmacokinetics**

##### **4.1. Metabolism/metabolites**

No data available to us at this time.

##### **4.2. Absorption, distribution and excretion**

“In a 120-day oral toxicity study in rats no indication of accumulation was reported. The Panel concludes that no accumulation of ethylene-vinyl acetate copolymer wax in man is anticipated.”

As taken from EFSA, 2014.

##### **4.3. Interactions**

No data available to us at this time.

#### **5. Toxicity**

##### **5.1. Single dose toxicity**

No data available to us at this time.

##### **5.2. Repeated dose toxicity**

EVA was implanted subcutaneously in dogs for 1 year. A thin fibrous capsule consisting of fibrous cells with flat nuclei was formed around the substance, and fibroblasts were found sporadically in the internal layer of the capsule. No inflammatory reaction was found (Kojima 1975).

“An oral 120-day rat study on a polymeric additive with a vinyl acetate content of 12-13 % (w/w) and an ethylene content of 87-88 % (w/w) related to the ethylene vinyl acetate copolymer wax under evaluation (vinyl acetate content 6-15% and ethylene content 85-94 %) was performed in 1966. Data were reported on the following parameters: body weights, hematology, urinalysis, serum glutamate pyruvate transaminase levels, liver and kidney weights and histological examinations of

10 inner organs. No substance-related changes were observed in the above parameters in rats fed with a diet containing 50 000 or 100 000 mg/kg bw/day of the test substance compared to control rats (approx. 4 000 or 8 000 mg/kg bw/day), i.e. the NOAEL in this study was considered to be 8 000 mg/kg bw/day ( the highest dose tested). Notwithstanding the limited end points of this study these data indicate that ethylene-vinyl acetate copolymer wax has a low sub-chronic toxicity. The LMWF content of that polymeric additive is not known, but given the similarity between the types of polymer, the content is anticipated to be similar to the LMWF content in the ethylene vinyl acetate copolymer under evaluation (i.e. approximately 10%). Nevertheless considering, conservatively, a LMWF content of only 1% in the polymeric additive used in the 120-day study, for the LMWF of that additive an NOAEL of 80 mg/kg bw/d can be estimated, (assuming that the more heavy fraction is completely non-toxic). The Panel concluded that this NOAEL value would provide a sufficiently large margin of safety of approximately 1000 compared to the exposure to the LMWF from the ethylene vinyl acetate copolymer under evaluation at a maximum migration level of 5 mg/kg food, bearing in mind that uncertainty due to read-across should also be taken into account. This conclusion is in line with supportive data from subchronic studies on oxidised polyethylene waxes evaluated by EFSA (EFSA, 2009). These data included five 90-day rat studies each on different commercial products, with the lowest NOAEL of 500 mg/kg bw/day and a reproduction and developmental toxicity diet rat study (OECD 421) on a LMWF of oxidised polyethylene waxes with a NOAEL of 1 000 mg/kg bw/day or higher.”

As taken from EFSA, 2014.

### *5.3. Reproduction toxicity*

**Development of a polymeric releasing device for 2'-carbomethoxyphenyl 4-guanidinobenzoate (a proteinase inhibitor): release rate, in vitro antifibrinolytic activity and in utero contraceptive effect (Abstract).** A polymeric delivery system consisting of ethylene-vinyl acetate copolymer (EVAc) was developed for 2'-carbomethoxyphenyl 4-guanidinobenzoate (MSGB), a potent inhibitor of the sperm enzyme acrosin. The optimal device consists of copolymer with 40% vinyl acetate by weight (EVAc/40), 65% drug loading and MSGB with a particle size of 250-499 micron. This formulation yields a device that is highly flexible and can be shaped to many forms and sizes. Construction of the device does not alter the properties of MSGB. Well controlled release of MSGB from the device occurs in vitro and in the uteri of rats. The in vitro release rate under "infinite sink" conditions is essentially the same as the in vivo release rate. The contraceptive effect of the MSGB-releasing device was tested in rabbits by placing a blank (control) device in one uterine horn and an MSGB-releasing device in the contralateral horn. In contrast to blank devices, MSGB-releasing devices completely prevent pregnancy, not only by inhibiting fertilization but also by decreasing implantation. MSGB possesses high in vitro antifibrinolytic activity. These results indicate that a very flexible device can be constructed for uterine application which retains its contraceptive effect by release of MSGB. The antifibrinolytic activity of MSGB may further decrease the menorrhagia that can be associated with IUD use.

As taken from Burns JW et al . Contraception. 1988, Sep; 38(3):349-64. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/3168452?dopt=AbstractPlus>

### *5.4. Mutagenicity*

“The acetate groups in ethylene-vinyl acetate copolymer wax are expected to be hydrolysed by esterases, similarly to vinyl acetate, and this would leave an ethylene-vinyl alcohol copolymer chain. Neither the copolymer itself nor its hydrolysis product have structural alerts for genotoxicity. Overall, the Panel concluded that the ethylene-vinyl acetate copolymer wax does not raise concern for genotoxicity.”

As taken from EFSA, 2014.



### *5.5. Cytotoxicity*

No data available to us at this time.

### *5.6. Carcinogenicity*

No data available to us at this time.

### *5.7. Irritation/immunotoxicity*

EVA was implanted subcutaneously in dogs for 1 year. A thin fibrous capsule consisting of fibrous cells with flat nuclei was formed around the substance, and fibroblasts were found sporadically in the internal layer of the capsule. No inflammatory reaction was found (Kojima 1975).

Elvax 40P (EVX), an ethylene vinyl-acetate copolymer, has been well characterised as an implant material that causes no inflammatory response and is capable of the sustained local release of a wide variety of undenatured macromolecules in vivo (Silberstein & Daniel 1982).

The vaginal immune response following controlled, local administration of a model antigen, ferritin, was determined by using ferritin-releasing ethylene-vinyl acetate copolymer (poly(ethylene-co-vinyl acetate)) vaginal rings to provide long term continuous antigen exposure in mice primed with subcutaneous (SC) or oral ferritin. SC primed mice receiving ferritin-loaded vaginal rings had ferritin-specific IgA in their mucus secretions, while mice receiving blank rings did not. Oral priming with ferritin-loaded polylactic acid (poly(lactic acid)) microspheres also produced significant levels of ferritin-specific IgA in the vaginal secretions, but required the presence of cholera toxin. It was concluded that controlled ferritin delivery to mucosal surfaces, either by oral, biodegradable microspheres or vaginal rings, provided a convenient and reliable method for enhancing vaginal IgA production in mice.

As taken from Wyatt TL et al. Controlled Release; VOL 50 ISS Jan 2 1998, P93-102.

A new method for local delivery of anti-adhesion monoclonal antibodies (MAbs) to an exposed mucosal surface was developed using ethylene-vinyl acetate copolymer (poly(ethylene-co-vinyl acetate)) controlled-release devices and their prophylactic potential was evaluated by examining leukocyte adhesion to apical surfaces of T84 cells in the presence of MAbs to leukocyte surface proteins. MAbs against the MAC-1 adhesion receptor inhibited neutrophil attachment to T84 cells by as much as 97%. MAbs against murine leukocyte receptors were produced from several hybridomas and incorporated into the devices. During incubation in sodium chloride (saline) buffer, small polymer discs continuously released active MAbs for 10 days. After insertion into vaginal canals of mice, these polymer disks produced high levels of anti-MAC-1 MAb for several days. It was concluded that MAbs against leukocyte adhesion molecules significantly inhibit the ability of leukocytes to interact with mucosal epithelia in vitro and that these same MAbs can be delivered directly to mucosal surfaces in an active form using polymeric controlled-release devices.

As taken from Parkhurst MR and Saltzman WM. J. Controlled Release; VOL 42 ISS Dec 1996, P273-288.

### *5.8. All other relevant types of toxicity*

**Release rates in rats of a macromolecule from an ethylene-vinyl acetate copolymer were shown to be indistinguishable from those of identical implants tested in vitro (Abstract).** Ethylene is a low molecular weight hydrocarbon gas with few toxicological properties. In sufficient concentrations ethylene depletes the oxygen level of air and through this mechanism acts as an asphyxiant. No long term toxicological problems have been attributed directly to the gas. Ethylene does not have locally toxic effects (Doull et al 1980). Due to the paucity of information regarding

ethylene-vinyl acetate copolymer, the remainder of the toxicological information presented concerns vinyl acetate.

As taken from Brown LR et al. J. Pharm. Sci.; VOL 72 ISS Oct 1983, P1181-1185.

**Toxicology of polymers for implant contraceptives for women (Abstract).** This article reviews the toxicology of polymers that are used in contraceptive implants. The two main classes of synthetic, nondegradable polymers used in the delivery of female contraceptives are silicone elastomers (e.g., Silastic) and ethylene co-vinyl acetate (EVA; ELVAX). The controversies surrounding the silicone breast implants have prompted several studies to evaluate the toxicity of silicones. The epidemiologic data obtained thus far have overwhelmingly concluded that no correlation exists between certain chronic symptoms, such as arthritis, in patients and silicone prosthesis. This conclusion has been echoed by the expert panel report by the Institutes of Medicine. Although the IOM report focused on the safety of silicone breast prosthesis, data emerging from the joint reconstruction area also suggests that Silastic is safe for in vivo use. The toxicological studies on EVA are few, and the conclusion thus far is that they elicit no adverse local or systemic response over extended periods in vivo. In conclusion, the prognosis for Silastic and ELVAX as of now is excellent. However, any future implant development using these polymers should place an emphasis on processing parameters to minimize potential small molecule leachants and establish safety as a function of both site and duration of implantation.

As taken from Shastri PV. Contraception. 2002, Jan; 65(1):9-13. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/11861050?dopt=AbstractPlus>

**PGE2 and angiogenesis (Abstract).** The angiogenic capability of PGE2 was tested by implanting pellets of an ethylene vinyl acetate slow release polymer containing PGE2 on the chorioallantoic membrane of 8-day-old chicken embryos. Elvax pellets releasing approximately 0.2, 2.0, or 20 ng/day PGE2 were found to induce neovascular responses. In contrast, pellets releasing 2.0 or 20 ng/day of either PGA2, PGF2, or TXB2 did not appear to be angiogenic when compared with PGE2. These release rates of PGE2 are similar to those reported for a variety of tumors, activated macrophages, inflammatory exudates, and rheumatoid synovia, suggesting that PGE2 may be a key factor in various neovascular reactions. As taken from DM; Auerbach R. Proc Soc Exp Biol Med. 1983, Feb; 172(2):214-8. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/6572402?dopt=AbstractPlus>

**Effect of PEG6000 on the in vitro and in vivo transdermal permeation of ondansetron hydrochloride from EVA1802 membranes (Abstract).** The objective was to evaluate ethylene vinyl acetate (EVA) copolymer membranes with vinyl acetate content of 18% w/w (EVA1802) for transdermal delivery of ondansetron hydrochloride. The EVA1802 membranes containing selected concentrations (0, 5, 10 and 15% w/w) of PEG6000 were prepared, and subjected to in vitro permeation studies from a nerodilol-based drug reservoir. Flux of ondansetron from EVA1802 membranes without PEG6000 was 64.1 +/- 0.6 microg/cm(2).h, and with 10%w/w of PEG6000 (EVA1802-PEG6000-10) it increased to 194.9 +/- 4.6 microg/cm(2).h. However, with 15%w/w of PEG6000, EVA1802 membranes produced a burst release of drug which in turn decreased drug flux. The EVA1802-PEG6000-10 membrane was coated with an adhesive emulsion, applied to rat epidermis and subjected to in vitro permeation studies against controls. Flux of ondansetron from transdermal patch across rat epidermis was 111.7 +/- 1.3 microg/cm(2).h, which is about 1.3 times the required flux. A TTS was fabricated using adhesive-coated EVA1802-PEG6000-10 membrane and other TTS components, and subjected to in vivo delivery in human volunteers against a control. It was concluded from the comparative pharmacokinetic study that TTS of ondansetron, prepared with EVA1802-PEG6000-10 membrane, provided average steady-state plasma concentration on par with multiple-dosed oral tablets, but with a low percent of peak-to-trough fluctuation.

As taken from Krishnaiah YS et al. Pharm Dev Technol. 2009; 14(1):50-61. PubMed, 2010 available at

[http://www.ncbi.nlm.nih.gov/entrez/query.fcgi?cmd=retrieve&db=pubmed&list\\_uids=18819031&dopt=AbstractPlus](http://www.ncbi.nlm.nih.gov/entrez/query.fcgi?cmd=retrieve&db=pubmed&list_uids=18819031&dopt=AbstractPlus)

**In vivo biocompatibility of three potential intraperitoneal implants (abstract).** The intraperitoneal biocompatibility of PDMS, polyHEMA and pEVA was investigated in rats, rabbits and rhesus monkeys. No inflammation was evidenced by hematological analyses and measurement of inflammatory markers throughout the experiment and by post-mortem examination of the pelvic cavity. After 3 or 6 months, histological analysis revealed fibrous tissue encapsulating PDMS and PEVA implants in all species and polyHEMA implants in rabbits and monkeys. Calcium deposits were observed inside polyHEMA implants. The intraperitoneal biocompatibility of all 3 polymers makes them suitable for the design of drug delivery systems, which may be of great interest for pathologies confined to the pelvic cavity. As taken from Defrère et al. (2011). *Macromol. Biosci.* 11(10):1336-45. PubMed, 2012, available at <http://www.ncbi.nlm.nih.gov/pubmed/21823236>

## **6. Functional effects on**

### **6.1. Broncho/pulmonary system**

No data available to us at this time.

### **6.2. Cardiovascular system**

**Ethylene vinylacetate copolymer particles dissolved in polyvinyl alcohol (2,000-mer) solution as an embolic material for vascular anomalies. A preliminary study (Abstract).**

We have prepared a new material for embolisation: ethylene vinylacetate copolymer dissolved in polyvinyl alcohol. When in contact with blood, polyvinyl alcohol rapidly becomes a soft gel, which is accompanied by wedging of the ethylene vinylacetate copolymer. We analysed the histopathology of intra-arterial microemboli in rats, after intracarotid injection of this material. We confirmed that it was applicable to embolisation for neurosurgical treatment.

As taken from Kinoshita A et al. *Neuroradiology.* 1994; 36(1):65-8. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/8108003?dopt=Abstract>

**Prevention of experimental cerebral vasospasm by intracranial delivery of a nitric oxide donor from a controlled-release polymer: toxicity and efficacy studies in rabbits and rats (Abstract).**

**BACKGROUND AND PURPOSE:** A reduction in the local availability of nitric oxide (NO) may play a role in the etiology of chronic cerebral vasospasm after subarachnoid hemorrhage (SAH). We investigated the toxicity and efficacy of a locally delivered NO donor from a controlled-release polymer in preventing experimental cerebral vasospasm in rats and rabbits, respectively.

**METHODS:** Diethylenetriamine/NO (DETA/NO) was incorporated into controlled release ethylene-vinyl acetate (EVAc) polymers. Twenty-eight rats were used in a dose-escalation toxicity study to establish a maximally tolerated dose of DETA/NO-EVAc polymer. In the efficacy experiment, 20 rabbits were assigned to 4 experimental groups (n=5 per group): sham operation; SAH only; SAH+empty EVAc polymer; and SAH+DETA/NO-EVAc polymer. Treatment was initiated 30 minutes after blood deposition. Basilar artery lumen patency was assessed 72 hours after hemorrhage to evaluate the efficacy of DETA/NO in preventing cerebral vasospasm. **RESULTS:** In the toxicity study, a dose of 3.4 mg/kg was identified as the LD(20) (dose with 20% mortality during the study period) of this DETA/NO formulation. Brain histology revealed hemorrhage and ischemic changes at the implantation site associated with high concentrations of DETA/NO. In the efficacy study, treatment with DETA/NO-EVAc polymer resulted in a significant decrease in basilar artery vasospasm compared with no treatment (93.0+/-4.9% versus 71.4+/-11.9%; P=0.035) or compared with treatment with blank EVAc polymer (93.0+/-4.9% versus 73.2+/-6.4%; P=0.003). **CONCLUSIONS:** Local delivery of DETA/NO prevents vasospasm in the rabbit basilar artery. Local delivery of DETA/NO via polymers is a safe and effective strategy for preventing cerebral vasospasm after SAH in this model.

As taken from Gabikian P; Clatterbuck RE et al. 2002, Nov; 33(11):2681-6. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/12411661?dopt=AbstractPlus>

### 6.3. Nervous system

No data available to us at this time.

### 6.4. Other organ systems, dependent on the properties of the substance

No data available to us at this time.

## 7. Addiction

JTI is not aware of any information that demonstrates that this ingredient has any addictive effect.

## 8. Burnt ingredient toxicity

Endpoint	Tested level (ppm)	Reference
Smoke chemistry	-	JTI Internal Report
In vitro genotoxicity	-	JTI Internal Report
In vitro cytotoxicity	-	JTI Internal Report

In comparison with a CSC of a reference cigarette with sideseam adhesives/cigarette paper corresponding to representative specifications for the majority of commercial cigarettes no differences were observed either in the bacterial mutagenicity, cytotoxicity or mammalian cell genotoxicity of the smoke condensate prepared from cigarettes with sideseam adhesives/cigarette paper containing Ethylene Vinyl Acetate Copolymer at 1.886 mg/cig. The smoke chemistry data between test and reference cigarette revealed small changes towards both higher and lower yields per cigarette over all analytical groups. These differences were well within the variability of the analytical methods (JTI NTM Study Report(s)).

When EVA was added to the wrappers of experimental flue-cured tobacco samples, increases in HCN and aldehydes were observed. There was no change in the level of acrolein or formaldehyde (Anon 1986).

## 9. Heated/vapor emissions toxicity

Aerosol from heated tobacco stick(s) containing Ethylene-Vinyl Acetate Copolymer was tested in aerosol chemistry and a battery of in vitro test(s). Under the test conditions and within the sensitivity and specificity of the bioassay(s), the activity of the total particulate matter (TPM) and/or gas vapor phase (GVP) were not increased by the addition of this ingredient when compared to TPM and/or GVP from reference combustible cigarettes. The table below provides the highest tested level(s) and specific endpoint(s):

Endpoint	Tested level (mg/stick)	Reference
Aerosol chemistry	11.0	Labstat International Inc. (2020a) Labstat International Inc. (2021a)
In vitro genotoxicity	11.0	Labstat International Inc. (2020b) Labstat International Inc. (2021b)
In vitro cytotoxicity	11.0	Labstat International Inc. (2020b) Labstat International Inc. (2021b)

## 10. Ecotoxicity

### 10.1. Environmental fate

The Ecological Categorization Results from the Canadian Domestic Substances List simply state that acetic acid ethenyl ester, polymer with ethane (CAS RN 24937-78-8) is persistent in the environment.

Data accessed May 2017 on the OECD website: <http://webnet.oecd.org/CCRWeb/Search.aspx>

### WATER FATE

**Evaluation of poly(ethylene-co-vinyl acetate-co-carbon monoxide) and polydimethylsiloxane for equilibrium sampling of polar organic contaminants in water (Abstract).** The aim of the present study was to develop a passive absorptive equilibrium sampler that would enable the determination of the concentrations of polar organic compound (POC) in water more efficiently than existing techniques. To this end, a novel plastic material, poly(ethylene-co-vinyl acetate-co-carbon monoxide) (PEVAC), was evaluated and the results were compared with an existing silicone-based passive absorptive equilibrium device. Seven compounds (imidacloprid, carbendazim, metoprolol, atrazin, carbamazepine, diazinon, and chlorpyrifos), a mixture of pharmaceuticals, and pesticides with a logarithmic octanol-water partition coefficient ranging from 0.2 to 4.77 were selected as model substances for the experiments. The results showed that six of the seven selected POCs reached distribution equilibrium within 4 d in the two materials tested. A linear relation with a regression coefficient of more than 0.8906 between the established logarithmic absorbent-water partition coefficient and the calculated logarithmic dissociation partition coefficient of the selected compounds in the two polymers was observed. The correlation between these two coefficients was within one order of magnitude for the compounds that reached equilibrium in the two polymers, which demonstrates that both materials are suitable for mimicking biological uptake of POCs. The PEVAC material showed an enhanced sorption for all selected compounds compared to the silicone material and up to five times higher enrichment for the most polar compound. Fluorescence analysis of the sampler cross-section, following the uptake of fluoranthene, and proof that the sorption was independent of surface area variations demonstrated that the PEVAC polymer possessed absorptive rather than adsorptive enrichment of organic compounds.

As taken from Magnér JA et al. Environ Toxicol Chem. 2009, Sep; 28(9):1874-80. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/19938334?dopt=AbstractPlus>

### *10.2. Aquatic toxicity*

The Ecological Categorization Results from the Canadian Domestic Substances List simply state that acetic acid ethenyl ester, polymer with ethane (CAS RN 24937-78-8) is not inherently toxic to aquatic organisms, giving a pivotal value for inherent toxicity of 14 mg/L.

Data accessed May 2017 on the OECD website: <http://webnet.oecd.org/CCRWeb/Search.aspx>

### *10.3. Sediment toxicity*

No data available to us at this time.

### *10.4. Terrestrial toxicity*

No data available to us at this time.

### *10.5. All other relevant types of ecotoxicity*

The Ecological Categorization Results from the Canadian Domestic Substances List simply state that acetic acid ethenyl ester, polymer with ethane (CAS RN 24937-78-8) is not bioaccumulative in the environment.

Data accessed May 2017 on the OECD website: <http://webnet.oecd.org/CCRWeb/Search.aspx>

## **11. References**

- AICIS (2012). Australian Government Department of Health. Australian Industrial Chemicals Introduction Scheme. Inventory Multi-Tiered Assessment and Prioritisation (IMAP) Tier I. Health record for acetic acid, ethenyl ester, copolymer with ethene (CAS RN 24937-78-8). Dated 7 December 2012. Available at <https://www.industrialchemicals.gov.au/chemical-information/search-assessments?assessmentcasnumber=24937-78-8>
- AICIS (Undated). Australian Government Department of Health. Australian Inventory of Industrial Chemicals. Record for CAS 24937-78-8. Available at Anon (1986). EVA – A cigarette paper additive. Submission to the Independent Scientific Committee on Smoking and Health. July 1986. Master file 121.4.53, A-Z/Ethylene Vinyl Acetate/1985 Onwards/E10 <https://www.industrialchemicals.gov.au/chemicals/acetic-acid-ethenyl-ester-copolymer-ethene>
- Brown LR et al. (1983). J. Pharm. Sci.; VOL 72 ISS Oct 1983, P1181-1185.
- Burns JW et al. (1988). Development of a polymeric releasing device for 2'-carbomethoxyphenyl 4-guanidinobenzoate (a proteinase inhibitor): release rate, in vitro antifibrinolytic activity and in utero contraceptive effect. Contraception. 1988, Sep; 38(3):349-64. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/3168452?dopt=AbstractPlus>
- Commission (2005). Provisional list of monomers and additives notified to European Commission as substances which may be used in the manufacture of plastics or coatings intended to come into contact with foodstuffs. Synoptic Document (updated to June 2005). SANCO D3/AS D(2005). Available at [http://www.contactalimentaire.com/fileadmin/ImageFichier\\_Archive/contact\\_alimentaire/Fichiers\\_Documents/Avis\\_de\\_AESA/synoptic\\_doc\\_en\\_-\\_version\\_June\\_2005.pdf](http://www.contactalimentaire.com/fileadmin/ImageFichier_Archive/contact_alimentaire/Fichiers_Documents/Avis_de_AESA/synoptic_doc_en_-_version_June_2005.pdf)



- CosIng. Cosmetic substances and ingredients database. Record for ethylene/VA copolymer (CAS RN 24937-78-8). Undated.. Available at <https://ec.europa.eu/growth/tools-databases/cosing/>
- CPID (undated). Consumer Product Information Database. Record for ethylene/vinyl acetate co-polymer (CAS RN 24937-78-8). Accessed June 2021. Available at <https://www.whatsinproducts.com/>
- Defrère et al. (2011). In vivo biocompatibility of three potential intraperitoneal implants Macromol. Biosci. 11(10):1336-45. PubMed, 2012. available at <http://www.ncbi.nlm.nih.gov/pubmed/21823236>
- ECHA (2023). European Chemicals Agency. Classification and Labelling (C&L) Inventory database. Last updated 12 June 2023. Available at: <https://echa.europa.eu/information-on-chemicals/cl-inventory-database>
- ECHA (undated a). European Chemicals Agency. Information on Chemicals. Record for acetic acid, ethenyl ester, copolymer with ethene (CAS 24937-78-8; EC no. 429-840-1).. Available at: <https://echa.europa.eu/information-on-chemicals/registered-substances>
- ECHA (undated b). European Chemicals Agency. Information on Chemicals. Available at:
- EFSA (2014). EFSA CEF Panel (EFSA Panel on Food Contact Materials, Enzymes, Flavourings and Processing Aids). Scientific opinion on the safety assessment of the substance ethylene-vinyl acetate copolymer wax, CAS No 24937-78-8 for use in food contact materials. Question No EFSA-Q-2013-00282, adopted on 18 December 2013. EFSA Journal 12(2), 3555. Available at: <http://onlinelibrary.wiley.com/doi/10.2903/j.efsa.2014.3555/epdf>
- FDA (2023a). US Food and Drug Administration. Electronic Code of Federal Regulations (eCFR). Title 21. Last updated 24 May 2023. Available at <https://ecfr.federalregister.gov/>
- FDA (2023b). US Food and Drug Administration. Inactive Ingredient Database. Data through 1 January 2023. Available at <https://www.accessdata.fda.gov/scripts/cder/iig/index.cfm>
- FDA (2023c). US Food and Drug Administration. Inventory of Food Contact Substances Listed in 21 CFR. Last updated 27 February 2023. Available at: <https://www.cfsanappsexternal.fda.gov/scripts/fdcc/index.cfm?set=IndirectAdditives>
- Form DM and Auerbach R (1983). PGE2 and angiogenesis Proc Soc Exp Biol Med. 1983, Feb; 172(2):214-8. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/6572402?dopt=AbstractPlus>
- Gabikian P et al. (2002). Prevention of experimental cerebral vasospasm by intracranial delivery of a nitric oxide donor from a controlled-release polymer: toxicity and efficacy studies in rabbits and rats. Stroke. 2002, Nov; 33(11):2681-6. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/12411661?dopt=AbstractPlus>
- Haz-Map (2020). Record for ethylenevinylacetate copolymer (CAS RN 24937-78-8). Last updated 29 July 2020. Accessed June 2021. Available at <https://haz-map.com/>
- Health Canada (2021). Drugs and Health Products. Natural Health Products Ingredients Database. Records for ethylene/VA copolymer and ethylene vinyl acetate (both CAS RN 24937-78-8). Both last updated 4 May 2021. Accessed June 2021. Available at <http://webprod.hc-sc.gc.ca/nhpid-bdipsn/ingredReq.do?id=16136&lang=eng> and <http://webprod.hc-sc.gc.ca/nhpid-bdipsn/ingredReq.do?id=14526&lang=eng>
- HULL et al. Polymer degradation and stability ISSN 0141-3910 CODEN PDSTDW. Congrès, European Conference on Fire Retardant Polymers No8, Alessandria, ITALIE (06/2001) 2002, vol. 77, no 2 (169 p.) (11 ref.), pp. 235-242
- IFRA (undated). International Fragrance Association. IFRA Transparency List. Available at <https://ifrafragrance.org/priorities/ingredients/ifra-transparency-list>
- JTI Internal Report
- JTI NTM Study Report(s)

- Kinoshita A et al. (1994). Ethylene vinylacetate copolymer particles dissolved in polyvinyl alcohol (2,000-mer) solution as an embolic material for vascular anomalies. A preliminary study. *Neuroradiology*. 1994; 36(1):65-8. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/8108003?dopt=Abstract>
- Kojima K. (1975). Interaction Between Polymeric Materials and Tissue. *Biodeterioration of Polymeric Materials*. Bull. Tokyo Med. Dent. Univ., 22: 263-272
- Krishnaiah YS et al. (2009). Effect of PEG6000 on the in vitro and in vivo transdermal permeation of ondansetron hydrochloride from EVA1802 membranes. *Pharm Dev Technol*. 2009; 14(1):50-61. PubMed, 2010 available at [http://www.ncbi.nlm.nih.gov/entrez/query.fcgi?cmd=retrieve&db=pubmed&list\\_uids=18819031&dopt=AbstractPlus](http://www.ncbi.nlm.nih.gov/entrez/query.fcgi?cmd=retrieve&db=pubmed&list_uids=18819031&dopt=AbstractPlus)
- Labstat International Inc. (2020a) Characterization of Heat-not-Burn Emissions. Analytical Test Report(s).
- Labstat International Inc. (2020b) Determination of Mutagenic Response (Ames), Cytotoxic Response (NRU) and Genotoxic Response (ivMN) of Mainstream Aerosol Total Particulate Matter (TPM) and Mainstream Gas Vapor Phase (GVP) of Heat-not-burn Products. Biological Activity Test Report(s)
- Labstat International Inc. (2021a). Characterization of Heat-not-Burn Emissions. Analytical Test Report(s).
- Labstat International Inc. (2021b). Determination of Mutagenic Response (Ames), Cytotoxic Response (NRU) and Genotoxic Response (ivMN) of Mainstream Aerosol Total Particulate Matter (TPM) and Mainstream Gas Vapor Phase (GVP) of Heat-not-burn Products. Biological Activity Test Report(s).
- Liu Y et al. (2020). Highly efficient composite flame retardants for improving the flame retardancy, thermal stability, smoke suppression, and mechanical properties of EVA. *Materials (Basel)* 13(5), 1251. DOI: 10.3390/ma13051251. PubMed, 2020 available at <https://pubmed.ncbi.nlm.nih.gov/32164212/>
- Magnér JA et al (2009). Evaluation of poly(ethylene-co-vinyl acetate-co-carbon monoxide) and polydimethylsiloxane for equilibrium sampling of polar organic contaminants in water *Environ Toxicol Chem*. 2009, Sep; 28(9):1874-80. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/19938334?dopt=AbstractPlus>
- OECD. Organization for Economic Cooperation and Development. The Global Portal to Information on Chemical Substances (eChemPortal). Acetic acid ethenyl ester, polymer with ethene (CAS No. 24937-78-8). Accessed May 2017. Available at <http://webnet.oecd.org/CCRWeb/Search.aspx>
- Parkhurst MR and Saltzman WM (1996). *J. Controlled Release*; VOL 42 ISS Dec 1996, P273-288.
- PubChem (2023). Record for Ethene;ethenyl acetate (CAS RN 24937-78-8). Created 9 August 2005. Last modified 13 June 2023. Available at <https://pubchem.ncbi.nlm.nih.gov/compound/175988>
- Schwöpe AD et al. (1987). Migration of Irganox 1010 from ethylene-vinyl acetate films to foods and food-simulating liquids., *Food Chem Toxicol*. 1987, Apr; 25(4):327-30.). PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/3583159>
- Shastri PV (2002). Toxicology of polymers for implant contraceptives for women. *Contraception*. 2002, Jan; 65(1):9-13. PubMed, 2010 available at <http://www.ncbi.nlm.nih.gov/pubmed/11861050?dopt=AbstractPlus>
- Silberstein G. B. & Daniel C. W. (1982). Elvax 40P Implants: Sustained, Local Release of Bioactive Molecules Influencing Mammary Ductal Development. *Developmental Biology*, 93: 272-278.
- US EPA (2023). US Environmental Protection Agency. Electronic Code of Federal Regulations (eCFR). Title 40. Last updated 8 June 2023. Available at <https://ecfr.federalregister.gov/>



- US EPA InertFinder Database (2023). Last updated 24 May 2023. Available at <https://iaspub.epa.gov/apex/pesticides/f?p=INERTFINDER:1:0::NO:1>
- US EPA Substance Registry Services (SRS) – TSCA and CDR lists. Available at [https://sor.epa.gov/sor\\_internet/registry/substreg/LandingPage.do](https://sor.epa.gov/sor_internet/registry/substreg/LandingPage.do)
- Wyatt TL et al. (1998). Controlled Release; VOL 50 ISS Jan 2 1998, P93-102.

## ***12. Other information***

No data available to us at this time.

## ***13. Last audited***

June 2023

## SCIENTIFIC OPINION

### Scientific Opinion on the safety assessment of the substance ethylene-vinyl acetate copolymer wax, CAS No 24937-78-8 for use in food contact materials<sup>1</sup>

EFSA Panel on Food Contact Materials, Enzymes, Flavourings and Processing Aids (CEF)<sup>2,3</sup>

European Food Safety Authority (EFSA), Parma, Italy

#### ABSTRACT

This scientific opinion of the EFSA Panel on Food Contact Materials, Enzymes, Flavourings and Processing Aids deals with the safety assessment of the polymeric additive ethylene-vinyl acetate copolymer wax, CAS No 24937-78-8, FCM substance No 00969 for use as a dispersing agent, lubricant, pigment carrier, and/or a processing aid in the production of plastic materials made from polymers such as polyethylene (PE), polypropylene (PP) or polyethylene terephthalate (PET). Final articles are intended for repeated contact with all types of foodstuffs at any conditions of time and temperature. The copolymer has a weight-averaged molecular weight higher than 6 000 Da and the low molecular weight fraction (LMFW) below 1 000 Da was estimated to be below 10 % w/w. The copolymer starts decomposing at temperatures above 230 °C, which is above the maximum process temperature of PE and PP but it is below the maximum process temperature of PET. The Panel considered that in the absence of information on possible thermal decomposition products, the use of the substance in PET should be excluded. The specific migration of the LMFW from polyolefins was conservatively estimated to be up to approximately 5.8 mg/kg food. There is no evidence of genotoxicity of ethylene-vinyl acetate copolymer wax. A 120 day oral toxicity study in rats showed no indication of accumulation. Therefore, the CEF Panel concluded that the substance ethylene-vinyl acetate copolymer wax does not raise a safety concern for the consumer if it is used as additive up to 2 % in polyolefins and the migration of low molecular weight oligomeric fraction below 1 000 Da does not exceed 5 mg/kg food.

© European Food Safety Authority, 2014

#### KEY WORDS

ethylene-vinyl acetate copolymer wax, CAS No 24937-78-8, FCM substance No 00969, food contact materials, safety assessment, evaluation

<sup>1</sup> On request from the Bundesamt für Verbraucherschutz und Lebensmittelsicherheit, Germany, Question No EFSA-Q-2013-00282, adopted on 18 December 2013.

<sup>2</sup> Panel members: Ulla Beckman Sundh, Mona-Lise Binderup, Claudia Bolognesi, Leon Brimer, Laurence Castle, Alessandro Di Domenico, Karl-Heinz Engel, Roland Franz, Nathalie Gontard, Rainer Gürtler, Trine Husøy, Klaus-Dieter Jany, Martine Kolf-Clauw, Catherine Leclercq (until July 2013), Jean-Claude Lhuguenot (until November 2012), Wim Mennes, Maria Rosaria Milana, Iona Pratt, Kjetil Svensson, Maria de Fátima Tavares Poças, Fidel Toldrá and Detlef Wölflé. One member of the Panel did not participate in the discussion on the subject referred to above because of potential conflicts of interest identified in accordance with the EFSA policy on declarations of interests. Correspondence: [cef@efsa.europa.eu](mailto:cef@efsa.europa.eu)

<sup>3</sup> Acknowledgement: The Panel wishes to thank the members of the Working Group on Food contact materials: Mona-Lise Binderup, Laurence Castle, Riccardo Crebelli, Alessandro Di Domenico, Roland Franz, Nathalie Gontard, Ragna Hetland Bogen, Martine Kolf-Clauw, Eugenia Lampi, Maria Rosaria Milana, Maria de Fátima Tavares Poças, Philippe Saillard, Kjetil Svensson and Detlef Wölflé for the preparatory work on this scientific opinion.

Suggested citation: EFSA CEF Panel (EFSA Panel on Food Contact Materials, Enzymes, Flavourings and Processing Aids), 2014. Scientific Opinion on the safety assessment of the substance ethylene-vinyl acetate copolymer wax, CAS No 24937-78-8 for use in food contact materials. EFSA Journal 2014;12(2):3555, 9 pp. doi:10.2903/j.efsa.2014.3555

Available online: [www.efsa.europa.eu/efsajournal](http://www.efsa.europa.eu/efsajournal)

## SUMMARY

Within the general task of evaluating substances intended for use in materials in contact with food according to the Regulation (EC) No 1935/2004 of the European Parliament and of the Council of 27 October 2004 on materials and articles intended to come into contact with foodstuffs, the CEF Panel received a request from the Bundesamt für Verbraucherschutz und Lebensmittelsicherheit, Germany, for safety assessment of the substance ethylene-vinyl acetate copolymer wax following a corresponding application from the applicant BASF SE, Germany.

The safety assessment of ethylene-vinyl acetate copolymer wax with CAS No 24937-78-8 and the FCM substance No 00969 was requested for use as additive in polymers such as polyethylene (PE), polypropylene (PP), or polyethylene terephthalate (PET) at a maximum use level of 2 % (w/w) in plastics. Final articles are intended for repeated contact with all types of foodstuffs at any condition of time and temperature.

The substance has not been evaluated by the SCF or AFC/CEF Panels. However, the comonomers, ethylene (85-94 %) with FCM substance No 00125 and vinyl acetate (6-15 %) with FCM substance No 00231, used to manufacture the copolymer are authorised as monomers for food contact materials with a specific migration limit (SML) of 12 mg/kg for vinyl acetate and no SML for ethylene.

The copolymer has a weight-averaged molecular weight higher than 6 000 Da and the low molecular weight fraction (LMWF) below 1 000 Da was estimated to be below 10 % w/w. The copolymer starts decomposing at temperatures above 230 °C, which is above the maximum process temperature of polyethylene and polypropylene but it is below the maximum process temperature of PET. The Panel considered that in the absence of information on possible thermal decomposition products, the use of the substance in PET should be excluded. Specific migration of the LMWF was estimated for extruded films of low density polyethylene (LDPE) containing copolymer wax (ethylene-vinyl acetate copolymer with a maximum content of 20 % of vinyl acetate). Tests were performed as time-dependent migration experiments using food simulants such as 95 % ethanol and olive oil. Migration of the LMWF from a worst-case polymer (i.e. LDPE) containing the highest intended concentration of the copolymer additive (i.e. 2%) was conservatively estimated to be up to approximately 5.8 mg/kg.

The substance is a polymeric additive without any structural alert for genotoxicity and it is manufactured using authorised monomers. In a 120-day oral toxicity study in rats no indication of accumulation was reported. The Panel concludes that no accumulation of ethylene-vinyl acetate copolymer wax in man is anticipated.

The CEF Panel concluded that the substance ethylene-vinyl acetate copolymer wax does not represent a safety concern for the consumer if the substance is only to be used as an additive up to 2 % w/w in polyolefin materials and articles and the migration of low molecular weight oligomeric fraction below 1 000 Da does not exceed 5 mg/kg food.

## TABLE OF CONTENTS

Abstract .....	1
Summary .....	2
Table of contents .....	3
Background as provided by the legislation .....	4
Terms of reference as provided by the legislation.....	4
Assessment .....	5
1. Introduction .....	5
2. General information.....	5
3. Data available in the dossier used for this evaluation.....	5
4. Evaluation.....	6
4.1. Non-toxicological data.....	6
4.2. Toxicological data.....	6
Conclusions .....	8
Documentation provided to EFSA .....	8
References .....	8
Glossary and abbreviations .....	9

## BACKGROUND AS PROVIDED BY THE LEGISLATION

Before a substance is authorised to be used in food contact materials and is included in a positive list EFSA's opinion on its safety is required. This procedure has been established in Articles 8 and 9 of the Regulation (EC) No 1935/2004 of the European Parliament and of the Council of 27 October 2004 on materials and articles intended to come into contact with food<sup>4</sup>.

According to this procedure the industry submits applications to the Member States competent Authorities which in their turn transmit the applications to the EFSA for their evaluation. The application is supported by a technical dossier submitted by the industry following the SCF guidelines for the "presentation of an application for safety assessment of a substance to be used in food contact materials prior to its authorisation" (EC, 2001).

In this case, EFSA received an application from Bundesamt für Verbraucherschutz und Lebensmittelsicherheit, Germany, requesting the evaluation of the additive ethylene-vinyl acetate copolymer wax with the CAS No 24937-78-8 and the FCM substance No 00969.

## TERMS OF REFERENCE AS PROVIDED BY THE LEGISLATION

EFSA is required to carry out assessment on the risks originating from the migration into food of the additive ethylene-vinyl acetate copolymer wax, intended to be used as an additive in plastic materials (PE, PP and PET) for food contact articles and to deliver a scientific opinion according to Regulation (EC) No 1935/2004 the European Parliament and of the Council on materials and articles intended to come into contact with food.

---

<sup>4</sup> Commission Regulation (EC) No 1935/2004 of 27 October 2004 on materials and articles intended to come into contact with food and repealing Directives 80/590/EEC and 89/109/EEC. OJ C 117, 30.4.2004, p. 1.

## ASSESSMENT

### 1. Introduction

The European Food Safety Authority was asked by the Bundesamt für Verbraucherschutz und Lebensmittelsicherheit, Germany, to evaluate the safety of ethylene-vinyl acetate copolymer wax with CAS No 24937-78-8 and FCM substance No 00969. The request has been included in the EFSA's register of received questions under number EFSA-Q-2013-00282. The dossier was submitted by the applicant, BASF SE, Germany.

### 2. General information

According to the applicant, the substance ethylene-vinyl acetate copolymer wax is a polymeric additive intended to be used as a dispersing agent, lubricant, pigment carrier, and/or processing aid, in the production of plastic materials made from polymers such as polyethylene (PE), polypropylene (PP) or polyethylene terephthalate (PET). The additive is intended to be used up to a maximum level of 2 % w/w in plastics for the food contact article. Final articles are intended for repeated contact with all types of foodstuffs under any conditions of time and temperature.

The substance has not been evaluated previously by the SCF or the AFC/CEF Panels. However, the co-monomers, ethylene (85-94 %) with FCM substance number 00125 and vinyl acetate (6-15 %) with FCM substance number 00231, used to manufacture the copolymer are authorised as monomers for food contact materials<sup>5</sup> with a specific migration limit (SML) of 12 mg/kg food for vinyl acetate and no SML for ethylene.

### 3. Data available in the dossier used for this evaluation

The studies submitted for evaluation followed the SCF guidelines for the presentation of an application for safety assessment of a substance to be used in food contact materials prior to its authorisation (EC, 2001).

#### Non-toxicity data:

- Data on identity
- Data on physical and chemical properties
- Data on intended use and authorisation
- Data on migration of the substance
- Data on residual content of the substance
- Data on oligomers
- Data on identification, quantification and migration of a reaction product

#### Toxicity data:

- Bacterial gene mutation test on oxidised polyethylene waxes
- *In vitro* mammalian cell gene mutation test on oxidised polyethylene waxes
- *In vitro* mammalian chromosome aberration test on oxidised polyethylene waxes
- 120-day oral toxicity study in rats
- 90-day oral toxicity rat studies on oxidised polyethylene waxes

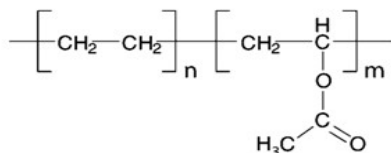
<sup>5</sup> Commission Regulation (EU) No 10/2011 of 14 January 2011 on plastic materials and articles intended to come into contact with food. Text with EEA relevance. OJ L 12, 15.1.2012, p. 1-89.

## 4. Evaluation

### 4.1. Non-toxicological data

Chemical formulae:  $[C_2H_4]_n[C_4H_6O_2]_m$

Chemical structure:



The copolymer has a weight averaged molecular weight (Mw) above 6 000 Da, a number averaged molecular weight (Mn) above 2 000 Da, and a molecular mass range of 200 - 10 000 Da. The copolymer is insoluble in water and in *n*-octanol. The Log Po/w was not provided. The low molecular weight fraction below 1 000 Da (LMWF) is estimated to be below 10 % w/w in the copolymer, therefore below 0.2 % (2000 mg/kg) in final articles.

The purity is higher than 99.98 %. Residual ethylene and vinyl acetate monomers in the copolymer are estimated to be no more than 50 and 150 mg/kg respectively. Ethylene monomer residues are unlikely to be present in the final article due to the chemical's volatility and the high temperatures and degassing steps used in manufacturing of the copolymer. The amount of vinyl acetate potentially migrating into food was calculated to be at least two orders of magnitude below the SML of 12 mg/kg.

The copolymer starts decomposing at temperatures above 230 °C, which is above the maximum process temperature of polyethylene and polypropylene but below the maximum process temperature of PET. Therefore, thermal decomposition of the copolymer may occur during the manufacture of PET materials and articles containing the additive. The possible degradation products were not addressed by the applicant. The Panel considered that in the absence of information on possible thermal decomposition products, the use of the substance in PET should be excluded. Specific migration of the LMWF was estimated for extruded films of low density polyethylene (LDPE) containing the copolymer (maximum content of 20 % of vinyl acetate). Tests were performed as time-dependent migration experiments using food simulants such as 95 % ethanol and olive oil. By using analysis of the simulants by gas chromatography and by gel permeation chromatography, along with migration modelling calculations, migration of the LMWF from a worst-case polymer (i.e. LDPE) containing the highest intended concentration of the copolymer (i.e. 2 %) was conservatively estimated to be up to approximately 5.8 mg/kg. Approximately 90 % of the migrated amount consisted of oligomers with a molecular weight below 500 Da and the remaining 10 % was in the range of 500 to 1 000 Da.

### 4.2. Toxicological data

The substance is a polymeric additive manufactured using the monomers ethylene and vinyl acetate, evaluated by the SCF in 1999 (SCF, 1999) and authorised with no SML and with a SML of 12 mg/kg food, respectively.

Ethylene is considered non-genotoxic and of low toxicological potential (HSBD, 2006).

A risk characterization of vinyl acetate was performed in the context of Council Regulation (EEC) No 793/93<sup>6</sup> on the evaluation and control of existing substances (EU-RAR, 2008). According to the EU-RAR, vinyl acetate is rapidly hydrolysed by carboxylesterases to acetic acid and vinyl alcohol, which quickly rearranges to acetaldehyde. At high concentrations of vinyl acetate the detoxifying activity of

<sup>6</sup> Council Regulation (EEC) No 793/93 of 23 March 1993 on the evaluation and control of the risks of existing substances. OJ L 84, 5.4.1993, p. 1.



aldehyde dehydrogenases is overwhelmed, and non-physiological high intracellular concentrations of acetaldehyde are produced. Acetaldehyde is a metabolic intermediate with low background concentrations, with genotoxic and carcinogenic effects limited to non-physiologically high concentrations. Genotoxicity data on vinyl acetate are in line with the hypothesis that vinyl acetate genotoxicity is mediated by acetaldehyde: similarly to acetaldehyde, vinyl acetate is genotoxic *in vitro*, with a threshold nonlinear dose-response (Budinsky et al., 2013), and non-genotoxic *in vivo* when evaluated at non-lethal doses (Albertini, 2013). The EU-RAR concluded that genotoxicity of vinyl acetate is based on a threshold mode of action, and it is limited to toxic doses (EU-RAR, 2008).

Long-term inhalation or oral administration of vinyl acetate to experimental animals produced tumors at the primary site of exposure. According to the EU-RAR, carcinogenicity of vinyl acetate is based on a secondary mechanism, due to the intracellular accumulation of acetaldehyde at high concentrations of vinyl acetate which results in increased cell proliferation and possibly DNA damage. Thus vinyl acetate is considered a high dose, threshold carcinogen (EU-RAR, 2008).

The Panel agreed with the conclusions of the EU-RAR and considered the low amounts of vinyl acetate possibly migrating into food to be of no toxicological concern.

The acetate groups in ethylene-vinyl acetate copolymer wax are expected to be hydrolysed by esterases, similarly to vinyl acetate, and this would leave an ethylene-vinyl alcohol copolymer chain. Neither the copolymer itself nor its hydrolysis product have structural alerts for genotoxicity. Overall, the Panel concluded that the ethylene-vinyl acetate copolymer wax does not raise concern for genotoxicity.

An oral 120-day rat study on a polymeric additive with a vinyl acetate content of 12-13 % (w/w) and an ethylene content of 87-88 % (w/w) related to the ethylene vinyl acetate copolymer wax under evaluation (vinyl acetate content 6-15% and ethylene content 85-94 %) was performed in 1966. Data were reported on the following parameters: body weights, hematology, urinalysis, serum glutamate pyruvate transaminase levels, liver and kidney weights and histological examinations of 10 inner organs. No substance-related changes were observed in the above parameters in rats fed with a diet containing 50 000 or 100 000 mg/kg bw/day of the test substance compared to control rats (approx. 4 000 or 8 000 mg/kg bw/day), i.e. the NOAEL in this study was considered to be 8 000 mg/kg bw/day (the highest dose tested). Notwithstanding the limited end points of this study these data indicate that ethylene-vinyl acetate copolymer wax has a low sub-chronic toxicity. The LMWF content of that polymeric additive is not known, but given the similarity between the types of polymer, the content is anticipated to be similar to the LMWF content in the ethylene vinyl acetate copolymer under evaluation (i.e. approximately 10%). Nevertheless considering, conservatively, a LMWF content of only 1% in the polymeric additive used in the 120-day study, for the LMWF of that additive an NOAEL of 80 mg/kg bw/d can be estimated, (assuming that the more heavy fraction is completely non-toxic). The Panel concluded that this NOAEL value would provide a sufficiently large margin of safety of approximately 1000 compared to the exposure to the LMWF from the ethylene vinyl acetate copolymer under evaluation at a maximum migration level of 5 mg/kg food, bearing in mind that uncertainty due to read-across should also be taken into account. This conclusion is in line with supportive data from subchronic studies on oxidised polyethylene waxes evaluated by EFSA (EFSA, 2009). These data included five 90-day rat studies each on different commercial products, with the lowest NOAEL of 500 mg/kg bw/day and a reproduction and developmental toxicity diet rat study (OECD 421) on a LMWF of oxidised polyethylene waxes with a NOAEL of 1 000 mg/kg bw/day or higher.

In the absence of data on potential accumulation and taking account of the likely hydrolysis of the substance to an ethylene-vinyl alcohol copolymer chain, the Panel considered as supporting evidence results from the studies on oxidised polyethylene waxes, which have a very low solubility in water and octanol, similar to ethylene-vinyl acetate copolymer wax. The results of these studies do not indicate an accumulation potential of oxidised polyethylene waxes, i.e. no precipitation was observed in



sensitive tissues such as liver or lymph nodes (EFSA, 2009). Therefore, the Panel concluded that similarly no accumulation of ethylene-vinyl acetate copolymer wax in man is anticipated.

## CONCLUSIONS

Having considered the above-mentioned data, the CEF Panel concluded that the substance ethylene-vinyl acetate copolymer wax does not raise a safety concern for the consumer if the substance is used as additive up to 2 % w/w in only polyolefin materials and articles and the migration of low molecular weight oligomeric fraction below 1 000 Da does not exceed 5 mg/kg food.

## DOCUMENTATION PROVIDED TO EFSA

1. Ethylene Vinyl Acetate, Copolymer Wax/eg A-C, Luwax. February 2013. Submitted by BASF SE.

## REFERENCES

- Albertini RJ, 2013. Vinyl acetate monomer (VAM) genotoxicity profile: relevance for carcinogenicity. *Critical Reviews in Toxicology*, 43, 8, 671-706.
- Budinsky R, Gollapudi B, Albertini R J, Valentine R, Stavanja M, Teeguarden J, Fensterheim R, Rick D, Lardie T, McFadden L, Green A and Recio L, 2013. Nonlinear responses for chromosome and gene level effects induced by vinyl acetate monomer and its metabolite, acetaldehyde in TK6 cells. *Environmental and Molecular Mutagenesis*, 54, 9, 755-768.
- EC (European Commission), 2001. Guidelines of the Scientific Committee on Food for the presentation of an application for safety assessment of a substance to be used in food contact materials prior its authorisation; [http://ec.europa.eu/food/fs/sc/scf/out82\\_en.pdf](http://ec.europa.eu/food/fs/sc/scf/out82_en.pdf)
- EFSA (European Food Safety Authority), 2009. Scientific Opinion of the Panel on food contact materials, enzymes, flavourings and processing aids (CEF) on a request related to a 23rd list of substances for food contact materials. *The EFSA journal* 2009, 1027-1029, 1-14.
- EU-RAR (European Union Risk Assessment Report), 2008. Vinyl acetate CAS No 108-05-4 EINECS No 203-545-4. Summary Risk Assessment Report. Final report, 2008. <http://echa.europa.eu/documents/10162/a3c24f78-4c8d-44e9-a424-24ac30c9c8aa>
- SCF (Scientific committee for food), 1999. Reports of the Scientific Committee for food (42nd series). Compilation of the evaluations of the scientific committee for food on certain monomers and additives used in the manufacture of plastics materials intended to come into contact with foodstuffs until 21 March 1997. 279pp. [http://ec.europa.eu/food/fs/sc/scf/reports/scf\\_reports\\_42.pdf](http://ec.europa.eu/food/fs/sc/scf/reports/scf_reports_42.pdf)
- HSBD (Hazardous Substances Data Bank) online, (last revision 2006/04/14). Ethylene. Available online: <http://www.toxnet.nlm.nih.gov/cgi-bin/sis/search/f?./temp/~ywjXpn:1>

**GLOSSARY AND ABBREVIATIONS**

AFC	Scientific Panel on Food Additives, Flavourings, Processing Aids and Materials in Contact with Food
CAS	Chemical Abstracts Service
CEF	Scientific Panel on Food Contact Materials, Enzymes, Flavourings and Processing Aids
Da	Dalton
EC	European Commission
EEC	European Economic Community
EFSA	European Food Safety Authority
EU	European Union
EU-RAR	European Union Risk Assessment Report
FCM	Food Contact Materials
LDPE	Low Density Polyethylene
LMWF	Low Molecular Weight Fraction
Mn	Averaged Molecular Weight
Mw	Molecular Weight
NOAEL	No Observed Adverse Effect Level
OECD	Organisation of Economic Co-operation and Development
SCF	Scientific Committee on Food
SML	Specific Migration Limit
PE	Polyethylene
PET	Polyethylene Terephthalate
Po/w	Octanol/Water partition coefficient
PP	Polypropylene

**SAFETY DATA SHEET**

according to Regulation (EC) No. 1907/2006

Version 6.4

Revision Date 15.04.2023

Print Date 08.05.2023

GENERIC EU MSDS - NO COUNTRY SPECIFIC DATA - NO OEL DATA

**SECTION 1: Identification of the substance/mixture and of the company/undertaking****1.1 Product identifiers**

Product name : Poly(ethylene-co-vinyl acetate)

Product Number : 437247

Brand : Aldrich

REACH No. : A registration number is not available for this substance as the substance or its uses are exempted from registration, the annual tonnage does not require a registration or the registration is envisaged for a later registration deadline.

CAS-No. : 24937-78-8

**1.2 Relevant identified uses of the substance or mixture and uses advised against**

Identified uses : Laboratory chemicals, Manufacture of substances

**1.3 Details of the supplier of the safety data sheet**

Company : Sigma-Aldrich Chemie GmbH  
Industriestrasse 25  
CH-9471 BUCHS

Telephone : +41 81 755 2511

Fax : +41 81 756 5449

E-mail address : technischerservice@merckgroup.com

**1.4 Emergency telephone**

Emergency Phone # : +41 43-508-2011 (CHEMTREC)  
+41 44-251-5151 (Tox-Zentrum)  
145(Tox Info Suisse)

**SECTION 2: Hazards identification****2.1 Classification of the substance or mixture**

Not a hazardous substance or mixture according to Regulation (EC) No 1272/2008.

**2.2 Label elements****Labelling according Regulation (EC) No 1272/2008**

Pictogram none

Signal Word none



Hazard statement(s)	none
Precautionary statement(s)	none
Supplemental Hazard Statements	none
EUH210	Safety data sheet available on request.

### 2.3 Other hazards

This substance/mixture contains no components considered to be either persistent, bioaccumulative and toxic (PBT), or very persistent and very bioaccumulative (vPvB) at levels of 0.1% or higher.

## SECTION 3: Composition/information on ingredients

### 3.1 Substances

Formula : C<sub>6</sub>H<sub>10</sub>O<sub>2</sub>  
CAS-No. : 24937-78-8

Component		Classification	Concentration
<b>vinyl acetate</b>			
CAS-No.	108-05-4	Flam. Liq. 2; Acute Tox. 4; Carc. 2; STOT SE 3; Aquatic Chronic 3; H225, H332, H351, H335, H412	>= 0,25 - < 1 %
EC-No.	203-545-4		
Index-No.	607-023-00-0		

For the full text of the H-Statements mentioned in this Section, see Section 16.

## SECTION 4: First aid measures

### 4.1 Description of first-aid measures

#### If inhaled

After inhalation: fresh air.

#### In case of skin contact

In case of skin contact: Take off immediately all contaminated clothing. Rinse skin with water/ shower.

#### In case of eye contact

After eye contact: rinse out with plenty of water. Remove contact lenses.

#### If swallowed

After swallowing: make victim drink water (two glasses at most). Consult doctor if feeling unwell.

### 4.2 Most important symptoms and effects, both acute and delayed

The most important known symptoms and effects are described in the labelling (see section 2.2) and/or in section 11

### 4.3 Indication of any immediate medical attention and special treatment needed

No data available



---

## SECTION 5: Firefighting measures

### 5.1 Extinguishing media

#### **Suitable extinguishing media**

Water Foam Carbon dioxide (CO<sub>2</sub>) Dry powder

#### **Unsuitable extinguishing media**

For this substance/mixture no limitations of extinguishing agents are given.

### 5.2 Special hazards arising from the substance or mixture

Carbon oxides

Combustible.

Development of hazardous combustion gases or vapours possible in the event of fire.

### 5.3 Advice for firefighters

In the event of fire, wear self-contained breathing apparatus.

### 5.4 Further information

Prevent fire extinguishing water from contaminating surface water or the ground water system.

---

## SECTION 6: Accidental release measures

### 6.1 Personal precautions, protective equipment and emergency procedures

Advice for non-emergency personnel: Avoid inhalation of dusts. Ensure adequate ventilation. Evacuate the danger area, observe emergency procedures, consult an expert. For personal protection see section 8.

### 6.2 Environmental precautions

Do not let product enter drains.

### 6.3 Methods and materials for containment and cleaning up

Cover drains. Collect, bind, and pump off spills. Observe possible material restrictions (see sections 7 and 10). Take up dry. Dispose of properly. Clean up affected area. Avoid generation of dusts.

### 6.4 Reference to other sections

For disposal see section 13.

---

## SECTION 7: Handling and storage

### 7.1 Precautions for safe handling

For precautions see section 2.2.

### 7.2 Conditions for safe storage, including any incompatibilities

#### **Storage conditions**

Tightly closed. Dry.

#### **Storage class**

Storage class (TRGS 510): 11: Combustible Solids

### 7.3 Specific end use(s)

Apart from the uses mentioned in section 1.2 no other specific uses are stipulated



---

## **SECTION 8: Exposure controls/personal protection**

### **8.1 Control parameters**

#### **Ingredients with workplace control parameters**

### **8.2 Exposure controls**

#### **Personal protective equipment**

##### **Eye/face protection**

Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU). Safety glasses

##### **Skin protection**

This recommendation applies only to the product stated in the safety data sheet, supplied by us and for the designated use. When dissolving in or mixing with other substances and under conditions deviating from those stated in EN374 please contact the supplier of CE-approved gloves (e.g. KCL GmbH, D-36124 Eichenzell, Internet: [www.kcl.de](http://www.kcl.de)).

Full contact

Material: Nitrile rubber

Minimum layer thickness: 0,11 mm

Break through time: 480 min

Material tested: KCL 741 Dermatril® L

Splash contact

Material: Nitrile rubber

Minimum layer thickness: 0,11 mm

Break through time: 480 min

Material tested: KCL 741 Dermatril® L

##### **Respiratory protection**

required when dusts are generated.

Our recommendations on filtering respiratory protection are based on the following standards: DIN EN 143, DIN 14387 and other accompanying standards relating to the used respiratory protection system.

Recommended Filter type: Filter type P2

The entrepreneur has to ensure that maintenance, cleaning and testing of respiratory protective devices are carried out according to the instructions of the producer. These measures have to be properly documented.

##### **Control of environmental exposure**

Do not let product enter drains.

---

## **SECTION 9: Physical and chemical properties**

### **9.1 Information on basic physical and chemical properties**

- a) Physical state                      Beads

Aldrich- 437247

Page 4 of 10

The life science business of Merck operates as MilliporeSigma in the US and Canada



b) Color	No data available
c) Odor	No data available
d) Melting point/freezing point	Melting point/range: 95 °C
e) Initial boiling point and boiling range	No data available
f) Flammability (solid, gas)	No data available
g) Upper/lower flammability or explosive limits	No data available
h) Flash point	No data available
i) Autoignition temperature	340 °C
j) Decomposition temperature	No data available
k) pH	No data available
l) Viscosity	Viscosity, kinematic: No data available Viscosity, dynamic: No data available
m) Water solubility	No data available
n) Partition coefficient: n-octanol/water	No data available
o) Vapor pressure	No data available
p) Density	0,933 g/cm <sup>3</sup> at 25 °C
Relative density	No data available
q) Relative vapor density	No data available
r) Particle characteristics	No data available
s) Explosive properties	No data available
t) Oxidizing properties	none

## 9.2 Other safety information

No data available

---

## SECTION 10: Stability and reactivity

### 10.1 Reactivity

The following applies in general to flammable organic substances and mixtures: in correspondingly fine distribution, when whirled up a dust explosion potential may generally be assumed.



## 10.2 Chemical stability

The product is chemically stable under standard ambient conditions (room temperature) .

## 10.3 Possibility of hazardous reactions

No data available

## 10.4 Conditions to avoid

no information available

## 10.5 Incompatible materials

Strong oxidizing agents, Strong acids

## 10.6 Hazardous decomposition products

In the event of fire: see section 5

---

## SECTION 11: Toxicological information

### 11.1 Information on toxicological effects

#### Acute toxicity

Oral: No data available

Inhalation: No data available

Dermal: No data available

#### Skin corrosion/irritation

Remarks: No data available

#### Serious eye damage/eye irritation

Remarks: No data available

#### Respiratory or skin sensitization

No data available

#### Germ cell mutagenicity

No data available

#### Carcinogenicity

No data available

#### Reproductive toxicity

No data available

#### Specific target organ toxicity - single exposure

No data available

#### Specific target organ toxicity - repeated exposure

No data available

#### Aspiration hazard

No data available

### 11.2 Additional Information

#### Endocrine disrupting properties

##### Product:

Assessment

The substance/mixture does not contain components considered to have endocrine disrupting properties according to REACH Article





57(f) or Commission Delegated regulation (EU)  
2017/2100 or Commission Regulation (EU)  
2018/605 at levels of 0.1% or higher.

To the best of our knowledge, the chemical, physical, and toxicological properties have not been thoroughly investigated.

---

## SECTION 12: Ecological information

### 12.1 Toxicity

No data available

### 12.2 Persistence and degradability

No data available

### 12.3 Bioaccumulative potential

No data available

### 12.4 Mobility in soil

No data available

### 12.5 Results of PBT and vPvB assessment

This substance/mixture contains no components considered to be either persistent, bioaccumulative and toxic (PBT), or very persistent and very bioaccumulative (vPvB) at levels of 0.1% or higher.

### 12.6 Endocrine disrupting properties

#### Product:

Assessment : The substance/mixture does not contain components considered to have endocrine disrupting properties according to REACH Article 57(f) or Commission Delegated regulation (EU) 2017/2100 or Commission Regulation (EU) 2018/605 at levels of 0.1% or higher.

### 12.7 Other adverse effects

No data available

---

## SECTION 13: Disposal considerations

### 13.1 Waste treatment methods

#### **Product**

See [www.retrologistik.com](http://www.retrologistik.com) for processes regarding the return of chemicals and containers, or contact us there if you have further questions.

---

## SECTION 14: Transport information

### 14.1 UN number

ADR/RID: -

IMDG: -

IATA: -

### 14.2 UN proper shipping name

ADR/RID: Not dangerous goods



IMDG:	Not dangerous goods		
IATA:	Not dangerous goods		
<b>14.3 Transport hazard class(es)</b>			
ADR/RID:	-	IMDG:	- IATA: -
<b>14.4 Packaging group</b>			
ADR/RID:	-	IMDG:	- IATA: -
<b>14.5 Environmental hazards</b>			
ADR/RID:	no	IMDG Marine pollutant:	no IATA: no
<b>14.6 Special precautions for user</b>			
No data available			
<b>Further information</b>			
Not classified as dangerous in the meaning of transport regulations.			

### 15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

**Authorisations and/or restrictions on use**

For this product a chemical safety assessment was not carried out

---

## SECTION 16: Other information

### Full text of other abbreviations

ADN - European Agreement concerning the International Carriage of Dangerous Goods by Inland Waterways; ADR - Agreement concerning the International Carriage of Dangerous Goods by Road; AIIC - Australian Inventory of Industrial Chemicals; ASTM - American Society for the Testing of Materials; bw - Body weight; CMR - Carcinogen, Mutagen or Reproductive Toxicant; DIN - Standard of the German Institute for Standardisation; DSL - Domestic Substances List (Canada); ECx - Concentration associated with x% response; ELx - Loading rate associated with x% response; EmS - Emergency Schedule; ENCS - Existing and New Chemical Substances (Japan); ErCx - Concentration associated with x% growth rate response; GHS - Globally Harmonized System; GLP - Good Laboratory Practice; IARC - International Agency for Research on Cancer; IATA - International Air Transport Association; IBC - International Code for the Construction and Equipment of Ships carrying Dangerous Chemicals in Bulk; IC50 - Half maximal inhibitory concentration; ICAO - International Civil Aviation Organization; IECSC - Inventory of Existing Chemical Substances in China; IMDG - International Maritime Dangerous Goods; IMO - International Maritime Organization; ISHL - Industrial Safety and Health Law (Japan); ISO - International Organisation for Standardization; KECI - Korea Existing Chemicals Inventory; LC50 - Lethal Concentration to 50 % of a test population; LD50 - Lethal Dose to 50% of a test population (Median Lethal Dose); MARPOL - International Convention for the Prevention of Pollution from Ships; n.o.s. - Not Otherwise Specified; NO(A)EC - No Observed (Adverse) Effect Concentration; NO(A)EL - No Observed (Adverse) Effect Level; NOELR - No Observable Effect Loading Rate; NZIoC - New Zealand Inventory of Chemicals; OECD - Organization for Economic Co-operation and Development; OPPTS - Office of Chemical Safety and Pollution Prevention; PBT - Persistent, Bioaccumulative and Toxic substance; PICCS - Philippines Inventory of Chemicals and Chemical Substances; (Q)SAR - (Quantitative) Structure Activity Relationship; REACH - Regulation (EC) No 1907/2006 of the European Parliament and of the Council concerning the Registration, Evaluation, Authorisation and Restriction of Chemicals; RID - Regulations concerning the International Carriage of Dangerous Goods by Rail; SADT - Self-Accelerating Decomposition Temperature; SDS - Safety Data Sheet; TCSI - Taiwan Chemical Substance Inventory; TECI - Thailand Existing Chemicals Inventory; TSCA - Toxic Substances Control Act (United States); UN - United Nations; UNRTDG - United Nations Recommendations on the Transport of Dangerous Goods; vPvB - Very Persistent and Very Bioaccumulative

### Further information

The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. Sigma-Aldrich Corporation and its Affiliates shall not be held liable for any damage resulting from handling or from contact with the above product. See [www.sigma-aldrich.com](http://www.sigma-aldrich.com) and/or the reverse side of invoice or packing slip for additional terms and conditions of sale.

Copyright 2020 Sigma-Aldrich Co. LLC. License granted to make unlimited paper copies for internal use only.

The branding on the header and/or footer of this document may temporarily not visually match the product purchased as we transition our branding. However, all of the



information in the document regarding the product remains unchanged and matches the product ordered. For further information please contact [mlsbranding@sial.com](mailto:mlsbranding@sial.com).

